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## Three new eudesmanoids from the Formosan soft coral Nephthea erecta

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# Three new eudesmanoids from the Formosan soft coral Nephthea erecta 

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#### Abstract

Chemical investigations on the organic extract of the Formosan soft coral Nephthea erecta led to the isolation of two new eudesmanoids, ( $4 R^{*}, 5 S^{*}, 6 \mathrm{Z}, 10 R^{*}$ )-8-oxo-eudesm-6-en-5 $\alpha, 11$-diol (1) and ( $6 E, 10 R$ )-4,5-dioxo-11-methoxy-eudesm-6-ene (2), together with one new tri-nor-eudesmane sesquiterpenoid, ( $4 S^{*}, 5 E, 10 R^{*}$ )-7-oxo-tri-nor-eudesm-5-en-4 3 -ol ( $\mathbf{3}$ ). The structures of metabolites $\mathbf{1 - 3}$ were elucidated through extensive spectroscopic analyses and by comparison with those reported in the literature. The anti-inflammatory activity using RAW 264.7 macrophages and their cytotoxicity against selected cancer cells of $\mathbf{1 - 3}$ were evaluated in vitro.


Keywords: Nephthea erecta; eudesmanoids; tri-nor-eudesmane; anti-inflammatory activity; RAW 264.7 macrophages; cytotoxicity

## 1. Introduction

Sesquiterpenoids and their analogs, especially those of the genus Nephthea, constitute a large family of secondary metabolites endowed with a range of structural diversity $[1-10]$. Some of these secondary metabolites have exhibited an array of biological activities such as insecticidal [2], cytotoxic [3-8], antiinflammatory [8,9], and antibacterial properties [8]. In the course of our ongoing endeavor to discover bioactive substances from marine organisms, chemical investigations of the Formosan soft coral Nephthea erecta Kükenthal were undertaken exhaustively. Two new eudesmanetype sesquiterpenoids, characterized as
$(4 R *, 5 S *, 6 Z, 10 R *)$-8-oxo-eudesm-6-en$5 \alpha, 11$-diol (1) and ( $6 E, 10 R$ )-4,5-dioxo-11-methoxy-eudesm-6-ene (2), and a new tri-nor-eudesmane sesquiterpenoid, ( $4 S^{*}$, $5 E, 10 R *$ )-7-oxo-tri-nor-eudesm-5-en-4 3 ol (3) (Figure 1), were isolated from the organism. The details of isolation and structural elucidation of these secondary metabolites are discussed in this paper. The in vitro anti-inflammatory activity using RAW 264.7 macrophages of $\mathbf{1}$ $(10 \mu \mathrm{M})$ exhibited the observed activity against iNOS protein expression, but no discernible activity against COX-2 protein expression. However, metabolites $\mathbf{2}$ and $\mathbf{3}$ inconspicuously reduced the levels of the iNOS and COX-2 proteins.

[^0]

1


2


3

Figure 1. Structures of metabolites $\mathbf{1 - 3}$.

## 2. Results and discussion

$(4 R *, 5 S *, 6 Z, 10 R *)$-8-Oxo-eudesm-6-en$5 \alpha, 11$-diol (1) was obtained as a colorless, viscous oil. The HR-ESI-MS of $\mathbf{1}$ exhibited a pseudomolecular ion peak at $\mathrm{m} / \mathrm{z}$ $275.1621[\mathrm{M}+\mathrm{Na}]^{+}$, consistent with the molecular formula of $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{3}$, requiring four degrees of unsaturation. The UV
spectrum showed the $\lambda_{\text {max }}(\mathrm{MeOH})$ value at 232 nm , indicating the presence of a cyclohex-2-enone moiety, as well as from a strong IR absorption at $1682 \mathrm{~cm}^{-1}$ (conjugated carbonyl group). The ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Table 1) of $\mathbf{1}$ also proved to contain resonances for the above functionality at $\delta_{\mathrm{H}} 6.86(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-6)$ and $\delta_{\mathrm{C}} 141.7$

Table 1. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopic data of $\mathbf{1 - 3}$. ${ }^{\text {a }}$

| C/H | 1 |  | 2 |  | 3 |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}$ | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}$ | ${ }^{13} \mathrm{C}$ | ${ }^{1} \mathrm{H}$ |
| 1 | $32.8(\mathrm{t})^{\text {b }}$ | $\begin{aligned} & \alpha: 1.86 \mathrm{dt}(13.2,4.4)^{\mathrm{c}} \\ & \beta: 1.17 \mathrm{br} \mathrm{~d}(13.2) \end{aligned}$ | 36.6 (t) ${ }^{\text {b }}$ | a: 1.56 m <br> b: 1.39 m | 41.2 (t) ${ }^{\text {b }}$ | $\begin{gathered} \alpha: 1.71 \mathrm{~m} \\ \beta: 1.34 \mathrm{td} \\ (13.2,3.2)^{c} \end{gathered}$ |
| 2 | 20.2 (t) | $\begin{aligned} & \alpha: 1.56 \mathrm{~m} \\ & \beta: 1.61 \mathrm{~m} \end{aligned}$ | 19.3 (t) | a: 1.52 m <br> b: 1.32 m | 17.3 (t) | $\begin{gathered} \alpha: 1.56 \mathrm{~m} \\ \beta: 2.10 \mathrm{dt} \\ (14.0,4.0) \end{gathered}$ |
| 3 | 29.9 (t) | $\begin{aligned} & \alpha: 1.61 \mathrm{~m} \\ & \beta: 1.53 \mathrm{~m} \end{aligned}$ | 44.6 (t) | $\begin{aligned} & \text { a: } 2.43 \mathrm{~m} \\ & \text { b: } 2.39 \mathrm{~m} \end{aligned}$ | 40.2 (t) | $\begin{aligned} & \alpha: 1.95 \mathrm{~m} \\ & \beta: 1.52 \mathrm{~m} \end{aligned}$ |
| 4 | 32.5 (d) | 1.94 m | 215.5 (s) |  | 71.5 (s) |  |
| 5 | 72.6 (s) |  | 207.8 (s) |  | 169.8 (s) |  |
| 6 | 141.7 (d) | 6.86 s | 123.5 (d) | 5.92 s | 123.2 (d) | 6.04 s |
| 7 | 144.3 (s) |  | 168.7 (s) |  | 201.2 (s) |  |
| 8 | 202.6 (s) |  | 25.7 (t) | 2.43 m | 34.1 (t) | $\begin{gathered} \alpha: 2.39 \mathrm{br} \mathrm{dt} \\ (17.6,3.6) \\ \beta: 2.61 \mathrm{ddd} \\ (17.6,14.4,4.8) \end{gathered}$ |
| 9 | 49.9 (t) | $\begin{aligned} & \alpha: 2.93 \mathrm{~d}(16.8) \\ & \beta: 2.05 \mathrm{~d}(16.8) \end{aligned}$ | 34.0 (t) | $\begin{aligned} & \alpha: 1.78 \mathrm{~m} \\ & \beta: 1.98 \mathrm{~m} \end{aligned}$ | 40.1 (t) | $\alpha: 1.74 \mathrm{~m}$ <br> $\beta: 1.86$ dd <br> (14.4, 4.8) |
| 10 | 40.6 (s) |  | 43.6 (s) |  | 35.8 (s) |  |
| 11 | 71.8 (s) |  | 73.6 (s) |  | 24.5 (q) | 1.46 s |
| 12 | 28.5 (q) | 1.42 s | 26.0 (q) | 1.34 s | 29.0 (q) | 1.42 s |
| 13 | 29.2 (q) | 1.43 s | 25.8 (q) | 1.35 s |  |  |
| 14 | 21.7 (q) | 1.05 s | 22.7 (q) | 1.08 s |  |  |
| 15 | 14.9 (q) | 1.00 d (6.4) | 30.5 (q) | 2.13 s |  |  |
| OMe |  |  | 51.4 (q) | 3.10 s |  |  |

[^1]

1


2


3

Figure 2. Selected $\mathrm{HMBC}(\rightarrow)$ and ${ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}$ COSY ( - ) correlations of $\mathbf{1}-\mathbf{3}$.
(CH, C-6), 144.3 (qC, C-7), and 202.6 (qC, C-8). In addition, its IR spectrum absorption band at $3439 \mathrm{~cm}^{-1}$ indicated the presence of a hydroxyl moiety. This assumption was further supported by the ${ }^{13} \mathrm{C}$ NMR signals resonating at $\delta_{\mathrm{C}} 72.6$ (qC, $\mathrm{C}-5$ ) and 71.8 (qC, C-11). The above functionalities accounted for two of the four degrees of unsaturation, indicating a bicyclic structure for 1.

From the COSY spectrum (Figure 2) of 1, it was possible to establish the proton sequence from $\mathrm{H}_{2}-1$ to $\mathrm{Me}-15$ through $\mathrm{H}_{2}-2, \mathrm{H}_{2}-3$, and $\mathrm{H}-4$. The cyclohex-2enone fragment was confirmed by the HMBC correlations from H-6 to C-8 and $\mathrm{C}-10$, and from $\mathrm{H}_{2}-9$ to $\mathrm{C}-5, \mathrm{C}-8$, and $\mathrm{C}-$ 10. Additionally, the HMBC spectrum exhibited correlations from $\mathrm{Me}-14$ to $\mathrm{C}-1$, $\mathrm{C}-5, \mathrm{C}-9$, and $\mathrm{C}-10$, from $\mathrm{Me}-15$ to $\mathrm{C}-3$ and C-4, and from H-6 to C-4 and C-5, as well as from $\mathrm{Me}-12$ and $\mathrm{Me}-13$ to $\mathrm{C}-11$ and C-7, establishing an eudesmane-type sesquiterpene skeleton of $\mathbf{1}$. The above HMBC correlations also led to the position of the two tertiary hydroxyl groups at C-5
and C-11, respectively. Thus, the planar structure of $\mathbf{1}$ was proposed decidedly.

The relative configuration of $\mathbf{1}$ was determined by the combination of the NOESY spectrum and a computer-generated lower energy conformation using MM2 force field calculations (Figure 3). The NOE correlations between $\mathrm{Me}-14$ with the following protons of $\mathrm{H}-1 \beta, \mathrm{H}-2 \beta$, $\mathrm{H}-4$, and $\mathrm{H}-9 \beta$ suggested $\beta$-orientation of the above protons, while the methyl group at $\mathrm{C}-4$ was oriented on the opposite side. $\mathrm{H}-2 \alpha$ was found to show NOE correlations with $\mathrm{H}-1 \alpha$ and $\mathrm{Me}-15$, indicating $\alpha$-orientation of Me-15. Moreover, the $\Delta$ value $\left(\delta_{\mathrm{CDCl}_{3}}-\delta_{\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}}\right)$ of the Me-14 protons $(400 \mathrm{MHz})$ of $\mathbf{1}$ in two different solvents $\left(\mathrm{CDCl}_{3}\right.$ and $\left.\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right)$ is small ( -0.06 ppm ), where the dihedral angle between $5-\mathrm{OH}$ and $\mathrm{Me}-14$ is large, indicating $\alpha$-orientation of 5-OH [11]. On the basis of the above observations, together with other detailed NOESY correlations, the structure of ( $4 R *, 5 S *, 6 Z, 10 R *$ )-8-oxo-eudesm-6-en- $5 \alpha, 11$-diol could be determined as shown in Figure 1.


Figure 3. Key NOESY correlations of $\mathbf{1}$ and $\mathbf{3}$.

The UV absorption maximum at 233 nm , and a strong IR absorption at $1684 \mathrm{~cm}^{-1}$, as well as the observation of the featuring carbon resonances at $\delta_{\mathrm{C}}$ 207.8 (qC, C-5), 123.5 (CH, C-6), and 168.7 (qC, C-7) in the ${ }^{13} \mathrm{C}$ NMR spectrum, revealed the presence of an $\alpha, \beta$-unsaturated carbonyl functionality in $\mathbf{2}$. Furthermore, a keto-carbonyl carbon was recognized as being present in $\mathbf{2}$ from its ${ }^{13} \mathrm{C}$ NMR signal at $\delta_{\mathrm{C}} 215.5$ (qC, C-4), as well as from a strong IR absorption at $1716 \mathrm{~cm}^{-1}$. The above functionalities accounted for three of the four degrees of unsaturation, suggesting a monocyclic skeleton for 2. By the assistance of extensive 2D NMR data, including COSY and HMBC correlations (Figure 2), the plane skeleton of $\mathbf{2}$ was fully proposed. Metabolite $\mathbf{2}$ is the methoxyl analog of chabrolidione B [7] and its HR-ESI-MS showed a pseudomolecular ion peak at $\mathrm{m} / \mathrm{z}$ $289.1779[\mathrm{M}+\mathrm{Na}]^{+}$, which is compatible with the molecular composition of $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{3}$. The ${ }^{1} \mathrm{H}$ NMR spectrum showed an additional proton singlet at $\delta_{\mathrm{H}} 3.10$ corresponding to $11-\mathrm{OMe}$. The ${ }^{13} \mathrm{C}$ NMR spectrum was comparable to that of chabrolidione B [7], except for an additional methoxyl signal at $\delta_{\mathrm{C}} 51.4$, which is located at $\mathrm{C}-11$ from the HMBC correlations from $\mathrm{Me}-12, \mathrm{Me}-13$, and 11OMe to C-11 (Figure 2). Furthermore, metabolite 2 has the same sign of optical rotation $\left([\alpha]_{D}-12\right)$ as that of chabrolidione $\mathrm{B}\left([\alpha]_{\mathrm{D}}-9.3\right)$. Thus, the absolute configuration of $\mathbf{2}$ was assumed to be $10 R$. On the basis of the above findings, the structure of $(6 E, 10 R)$-4,5-dioxo-11-meth-oxy-eudesm-6-ene (2) was established unambiguously.

Metabolite $\mathbf{3}$ was isolated as a colorless, viscous oil. According to HR-ESI-MS at $\mathrm{m} / \mathrm{z}$ $217.1206[\mathrm{M}+\mathrm{Na}]^{+}$and ${ }^{13} \mathrm{C}$ NMR spectral data, its molecular formula was established as $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{O}_{2}$, implying four degrees of unsaturation. The UV spectrum showed $\lambda_{\text {max }}(\mathrm{MeOH})$ value at 235 nm , indicating the presence of a cyclohex-2-enone
fragment, as well as from IR absorption at $1681 \mathrm{~cm}^{-1}$. In addition, the ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra (Table 1) of $\mathbf{3}$ further proved to contain resonances for a cyclohex-2enone moiety [ $\delta_{\mathrm{H}} 6.04(1 \mathrm{H}, \mathrm{s}, \mathrm{H}-6)$ and $\delta_{\mathrm{C}}$ 169.8 (qC, C-5), 123.2 (CH, C-6) and 201.2 (qC, C-7)]. Since the above functionality accounted for two of the four degrees of unsaturation, metabolite $\mathbf{3}$ was suggested to be a bicyclic framework. From the COSY spectrum of $\mathbf{3}$ (Figure 2), it was possible to establish the two partial structures of consecutive proton systems extending from $\mathrm{H}_{2}-1$ to $\mathrm{H}_{2}-3$ through $\mathrm{H}_{2}-2$, and from $\mathrm{H}_{2}-8$ to $\mathrm{H}_{2}-9$. The connectivities between $\mathrm{C}-1$ and $\mathrm{C}-10, \mathrm{C}-9$ and $\mathrm{C}-10, \mathrm{C}-5$ and $\mathrm{C}-10$ were confirmed by HMBC correlations of Me-11 with C-1, C-5, C-9, and C-10 (Figure 2). Furthermore, the HMBC correlations from $\mathrm{Me}-12$ to C-3, C-4, and C-5 suggested that C-3/C-4 and $\mathrm{C}-4 / \mathrm{C}-5$ were connected, and led the assignment of the tertiary hydroxyl group at C-4. The relative stereochemistry of $\mathbf{3}$ assigned by the NOESY spectrum was compatible with that of $\mathbf{3}$ obtained by computer modeling (Figure 3), in which the close contacts of atoms calculated in space were consistent with the NOESY correlations. The cross-peaks between Me11 and $\mathrm{H}-1 \beta, \mathrm{H}-2 \beta$, and $\mathrm{H}-8 \beta$ established the $\beta$-configuration of Me-11. Furthermore, the observed NOE interactions between $\mathrm{H}-3 \alpha / \mathrm{H}-1 \alpha$ and $\mathrm{H}-3 \alpha / \mathrm{Me}-12$ confirmed the $\alpha$-orientation of $\mathrm{Me}-12$ and supported the $\beta$-orientation of $4-\mathrm{OH}$. From the aforementioned results, metabolite 3 was fully formulated as ( $4 S *, 5 E, 10 R *$ )-7-oxo-tri-nor-eudesm-5-en-4 3 -ol.

The determination of absolute stereochemistry of $\mathbf{1}$ and $\mathbf{3}$ could not be carried out due to the absence of the secondary hydroxyls or the paucity of the material. It was noteworthy to mention that the absolute configurations of $\mathbf{1}$ and $\mathbf{3}$, as shown in their formulas, were confirmed after the determination of the absolute configuration of $\mathbf{2}$ on the basis of biosynthetic reasoning.

A




Figure 4. Effect of compounds $\mathbf{1 - 3}$ at $10 \mu \mathrm{M}$ on the LPS-induced pro-inflammatory iNOS and COX-2 protein expression of RAW 264.7 macrophages by immunoblot analysis. (A) Immunoblot of iNOS; (B) immunoblot of COX-2, and (C) immunoblot of $\beta$-actin. The values are mean $\pm$ SEM $(n=5)$. The relative intensity of the LPS-alone stimulated group was taken as $100 \%$. Significantly different from LPS-stimulated (control) group ( $* P<0.05$ ).

The in vitro anti-inflammatory activities of metabolites $\mathbf{1 - 3}$ were tested using LPS-stimulated cells. Stimulation of RAW 264.7 macrophage cells with LPS resulted in the upregulation of the pro-inflammatory iNOS and COX-2 proteins. At a concentration of $10 \mu \mathrm{M}$, metabolites $\mathbf{1 - 3}$ reduced the levels of the iNOS protein ( $79.5 \pm 7.1$, $99.1 \pm 5.8$, and $106.7 \pm 1.8 \%$, respectively) and COX-2 protein ( $94.2 \pm 4.3$, $101.9 \pm 3.3$, and $98.5 \pm 7.1 \%$, respectively), comparing with the control cells (LPS alone; Figure 4). The primary antiinflammatory results of $\mathbf{1}$ exhibited the observed activity against iNOS protein expression, but no discernible activity against COX-2 protein expression. In addition, metabolites 2 and 3 showed insignificant anti-inflammatory activity against LPS-stimulated RAW 264.7 macrophages. The in vitro anti-inflammatory assays were carried out according to the procedure described previously [12].

Metabolites 1-3 were tested for cytotoxicity against P-388 (mouse lymphocytic leukemia), HT-29 (human colon adenocarcinoma), and A549 (human lung carcinoma) cancer cell lines. The results showed that metabolites $\mathbf{1 - 3}$ were not cytotoxic to A549 and HT-29 cells. However, metabolites 1-3 exhibited significant cytotoxicity against P-388 cell line with $\mathrm{ED}_{50}$ values of $2.60,2.46$, and $2.42 \mu \mathrm{~g} / \mathrm{ml}$, respectively. The anticancer agent mithramycin was used as the
positive control and exhibited $\mathrm{ED}_{50}$ values of $0.06,0.08$, and $0.07 \mu \mathrm{~g} / \mathrm{ml}$ and against P-388, HT-29, and A549 cells, respectively. The experimental details of this assay were carried out according to a previously described procedure [13].

## 3. Experimental

### 3.1 General experimental procedures

Optical rotations were determined using a JASCO P1020 digital polarimeter. IR spectra were recorded on a JASCO FT/IR-4100 infrared spectrophotometer. UV spectra were obtained on a JASCO V650 spectrophotometer. The NMR spectra were recorded on a Varian MR 400 NMR spectrometer $\left(400 \mathrm{MHz}\right.$ for ${ }^{1} \mathrm{H}$ and 100 MHz for ${ }^{13} \mathrm{C}$ ), using $\mathrm{CDCl}_{3}$ with TMS as the internal standard. Chemical shifts are given in $\delta$ (ppm) and coupling constants in Hz. ESI-MS were recorded by ESI FT-MS on a Bruker APEX II mass spectrometer. Silica gel 60 (230-400 mesh; Merck, Darmstadt, Germany) and LiChroprep RP-18 (40-63 $\mu \mathrm{m}$; Merck) were used for column chromatography. Precoated silica gel plates (Kieselgel 60 $\mathrm{F}_{254}, 0.25 \mathrm{~mm}$; Merck) and precoated RP$18 \mathrm{~F}_{254 \mathrm{~s}}$ plates (Merck) were used for analytical TLC analyses. High-performance liquid chromatography (HPLC) was carried out using a Hitachi L-7100 pump equipped with a Hitachi L-7400 UV
detector at 220 nm and a semi-preparative reversed-phase column (Hibar Purospher RP-18e, $5 \mu \mathrm{~m}, 250 \mathrm{~mm} \times 10 \mathrm{~mm}$; Merck).

### 3.2 Animal material

The Formosan soft coral N. erecta was collected by hand using scuba at the Green Island located in the southeast coast of Taiwan, in July 2005, at a depth of 10 m , and was stored in a freezer for 5 weeks until extraction. This soft coral was identified by one of the authors (C.-F.D.). A voucher specimen (GN-80) has been deposited at the Department of Marine Biotechnology and Resources, National Sun Yat-sen University.

### 3.3 Extraction and isolation

A specimen of $N$. erecta was extracted repeatedly with fresh MeOH at room temperature. The combined MeOH extracts were evaporated under reduced pressure at $35^{\circ} \mathrm{C}$ to yield a brown viscous gum. The resulting MeOH extract ( 320 mg ) was subjected to column chromatography on silica gel using $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and MeOH gradient (100:1-0:1) for elution, to give 30 fractions. A fraction eluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}-\mathrm{MeOH}$ (80:1) was further purified by RP-HPLC using $65 \% \mathrm{MeOH}$ in $\mathrm{H}_{2} \mathrm{O}$ to afford $1(2 \mathrm{mg}), 2(1 \mathrm{mg})$, and $\mathbf{3}$ $(1 \mathrm{mg})$.

### 3.3.1 $\left(4 R^{*}, 5 S^{*}, 6 \mathrm{Z}, 10 R^{*}\right)-8-$ Oxo-eudesm-6-en-5 $\alpha$,11-diol (1)

Colorless, viscous oil; $[\alpha]_{\mathrm{D}}^{24}+52$ $\left(c=0.1, \mathrm{CHCl}_{3}\right) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\text {max }}$ $(\log \varepsilon): 232$ (3.85); IR (KBr) $\nu_{\max }: 3439$, 2932, 1682, 1660, 1461, 1378, 1165, 1126, $979,747 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopic data, see Table 1; ESI-MS m/z: 275
 $[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{15} \mathrm{H}_{24} \mathrm{O}_{3} \mathrm{Na}$, 275.1623). Selected ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{C}_{5} \mathrm{D}_{5} \mathrm{~N}\right.$, $400 \mathrm{MHz}) \delta 6.23(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{H}-6), 2.63(1 \mathrm{H}$, d, $J=16.8 \mathrm{~Hz}, \mathrm{H}-9 \alpha), 2.19(1 \mathrm{H}, \mathrm{d}$,
$J=16.8 \mathrm{~Hz}, \mathrm{H}-9 \beta), 2.16(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-4)$, $1.88(1 \mathrm{H}, \mathrm{m}, \mathrm{H}-1 \alpha), 1.72(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}-13)$, $1.64(3 \mathrm{H}, \mathrm{s}, \mathrm{Me}-12), 1.12(3 \mathrm{H}, \mathrm{d}, J=6.4$, Me-15), 1.11 (3H, $J=6.8$, Me-14).

### 3.3.2 (6E,10R)-4,5-Dioxo-11-methoxy-

 eudesm-6-ene (2)Colorless, viscous oil; $[\alpha]_{D}^{24}-12(c=0.1$, $\left.\mathrm{CHCl}_{3}\right) ; \mathrm{UV}(\mathrm{MeOH}) \lambda_{\max }(\log \varepsilon): 233$ (3.73); IR (KBr) $\nu_{\max }: 2927,1716,1684$, 1668, 1457, 1377, 1361, 1262, 1170, 1071, $802,749 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopic data, see Table 1; ESI-MS m/z: 289 [M +Na$]^{+}$; HR-ESI-MS m/z: 289.1779 $[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{16} \mathrm{H}_{26} \mathrm{O}_{3} \mathrm{Na}$, 289.1780).

### 3.3.3 ( $4 S^{*}, 5 E, 10 R *$ )-7-Oxo-tri-nor-eudesm-5-en-4 $\beta$-ol (3)

Colorless, viscous oil; $[\alpha]_{D}^{24}-6(c=0.1$, $\mathrm{CHCl}_{3}$ ); UV (MeOH) $\lambda_{\text {max }}(\log \varepsilon): 235$ (3.82); IR (KBr) $\nu_{\text {max }}: 3419,2926,1681$, 1666, 1458, 1375, 1209, 1124, $876 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectroscopic data, see Table 1; ESI-MS m/z: $217[\mathrm{M}+\mathrm{Na}]^{+}$; HR-ESI-MS m/z: 217.1206 $[\mathrm{M}+\mathrm{Na}]^{+}$(calcd for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{O}_{2} \mathrm{Na}$, 217.1204).

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[^1]:    ${ }^{\text {a }}$ Spectra were measured in $\mathrm{CDCl}_{3}\left({ }^{1} \mathrm{H}, 400 \mathrm{MHz}\right.$ and $\left.{ }^{13} \mathrm{C}, 100 \mathrm{MHz}\right)$.
    ${ }^{\mathrm{b}}$ Multiplicities are deduced by HSQC and DEPT experiments.
    ${ }^{c} J$ values (in Hz) are in parentheses.

